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SYNTHESIS AND PROPERTIES OF HYDROPHOBIZATORS BASED ON ACRYLATES AND METHACRYLATES OF POLYFLUOROALCOXY-1-ALKANOLES

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Abstract: Polyfluoroalkyl ethers of acrylic, methacrylic acids and polymers based on them are obtained. Hydrophobic properties of polymers are studied.

Keywords: Polyfluoroalkyl α -chloroethers, polyfluoroalkoxy ethylacrylates, methacrylates, tissue hydrophobization.

Earlier [1-5] α -chloroethers containing different structural fragments including polyfluoroalkyl substituents have been synthesized. High reactivity of chlorine atom located at α -position to electron-seeking polyfluoroalkyl group is used in the reactions of nucleophilic substitution.

Based on polyfluorinated telomer alccohols α -chloroethers have been obtained, they have been transformed into corresponding monomers using the reaction with potassium salts of acrylic and methacrylic acids:

$$\begin{split} H(CF_2CF_2)_nCH_2OH + RCHO + HCl &\rightarrow H(CF_2CF_2)_nCH_2OCH(R') - Cl \\ &\downarrow CH_2 = C(R')COONa \\ &H(CF_2CF_2)_nCH_2OCH(R)OC(O)C(R') = CH_2 \\ &n = 1 - 3; \ R; \ R' = H; \ CH_3 \end{split}$$

 α -Chloroethers based on polyfluorinated alcohols are synthesized in chloroform with the yield of up to 71% [1]. These are easy boiling liquids purified from admixtures by vacuum distillation.

The reaction of potassium salt of acrylic or metacrylic acid is carried out in dry chladone-113 at zero temperature with its further increasing up to 40 $^{\circ}$ C. After separation of laid-down salt (potassium chloride) filtrate is washed through with 5% solution of potassium bicarbonate, water till reaching the neutral reaction, dried with magnesium sulphate [6]. After isolation of solvent under vacuum the residue is distilled in vacuum. The yield of monomers for metacrylates is about 86,2-91,4 %, and for acrylates it is somewhat lower -- 76,0-85,4 %. Physical and chemical properties of monomers are listed in Table 1.

Table 1. Polyfluoroalcoxy ethylacrylates and Metacrylates, $H(CF_2CF_2)nCH_2OCH(CH_3)OC(O)C(R)=CH_2$

R	n	BP , °C/mm Hg	n ²⁰ D	d ²⁰ n
н	1	57-58/3	1,3746	1,2752
CH ₃	1	57-58/5	1,3816	1,2343
Н	2	79-80/5	1,3620	1,4236
CH ₃	2	71-78/4	1,3692	1,3712
Н	3	78-79/4	1,3515	1,5491
CH ₃	3	82-83/4	1,3559	1,4798

Polymers based on polyfluoroalcoxy ethylacrylates and methacrylates are being obtained by radical polymerization in the block for corresponding monomers at t of 75-80 °C in 1-1,5 hour. Hydrophobizing properties of solutions in chladone 113 have been being studied using glass fibre cloth KT-II-c 8/3-03-TO. Glass fibre cloth was being put into the solution of fluoropolymer, and afterwards upon drying the surplus weight and water absorption in % have been determined (Table 2).

Table 2. Hydrophobic Properties of Glass Fibre Cloth Grade KT-II-c 8/3-03-TO Processed With Solutions Polyfluoroalcoxy ethylmethacrylates, $H(CF_2CF_2)nCH_2OCH(CH_3)OC(O)C(CH_3)-CH_2-$.

Hydrophobizing (Water- repellent) Compound	Solution Concentration, %	Sample Gain Weight, %	Water Absorption
	2	1	34,3
	5	1,4	25,4
n = 1	8	2,2	24,8
	10	3,5	14,0
	2	1,2	24,3
	5	2,4	15,7
n = 2	8	3,2	14,8
	10	4,2	14,0
	2	1,3	19,1
	5	1,8	14,5
n = 3	8	2,9	13,7
	10	4,5	12,5
non-hydrophobized	-	-	66,1

cloth

As you can see from Table 2 the hydrophobization of cloth allows lowering the KT-II-c 8/3-03-TO glass fibre cloth's water absorption more than twice. Hydrophobic properties of glass fibre cloth 33/I-100 (VII-76) have been studied using solutions of lower concentration (Table 3).

Table 3. Hydrophobic Properties of Glass Fibre Cloth 93/I-100 (VII-76) Processed With Solutions Of Polymfluoroalkoxy ethylmethacrylates, $H(CF_2CF_2)nCH_2OCH(CH_3)OC(O)C(CH_3)-CH_2-$.

Hydrophobizing (Water-repellent) Compound	Solution Concentration, %	Sample Gain Weight, , %	Water Absorption	Capillarity, %
	0,5	0,3	27,4	80
	1	0,5	26,6	75
n = 1	2	0,8	25,8	70
	3	1,3	25,2	65
	0,5	0,4	26,4	72
n = 2	1	0,7	25,7	68
n = 2	2	1,2	24,3	65
	3	1,5	24,1	64
	0,5	0,5	25,2	70
n - 2	1	0,7	25,4	67
n = 3	2	1,3	24,0	67
	3	1,5	23,2	65
unhydrophobized cloth			28,9	90

Tensile strength of glass fibre cloth elemental threads (Table 4) processed with hydrophobizators is very important..

Table 4. Tensile strength of Glass Fibre Cloth Elemental Threads Processed With 0,5 % Solution of Polyfluoroalkoxy ethylmethacrylate At T 105 °C, Time Period – 20 min.

Serial #	Hydrophobized Cloth, Tensile Strength, $kg \cdot sec$	Serial #	Hydrophobized Cloth, Tensile Strength, <i>kg</i> · sec
1	0,95	1	0,75
2	0,99	2	0,85
3	0,95	3	0,88

	cp. 0,96		cp. 0,82
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As you can see from Table 4 the elemental threads of glass fibre cloth processed with the solution of polyfluoroalkoxy ethylmethacrylate are of somewhat bigger tensile strength than unhydrophobized cloth.

Thus, polymers based on polyfluoroalkoxy ethylacrylates and methacrylates can be recommended as hydrophobizators.

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