# Perfluorocyclobutene and its dimers in Diels-Alder reaction with cyclopentadiene and furan. New data 

A. Yu. Volkonskii,* E. M. Kagramanova, E. I. Mysov, N. D. Kagramanov

## A.N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences, ul. Vavilova 28, V-334, GSP-1, 119991 Moscow, Russia <br> e-mail: volkonskii@ineos.ac.ru


#### Abstract

Fluorinated norbornenes and 7-oxanorbornene, containing perfluorocyclobutane fragments, were synthesized by Diels-Alder reaction from perfluorocyclobutene or perfluorocyclobutene dimers and cyclopentadiene or furan.


Keywords: perfluorocyclobutene, perfluorocyclobutene dimers, cyclopentadiene, furan, Diels-Alder reaction.

Fluorinated polymers on the base of norbornene and its analogues were found to be promising materials for extreme UV lithography (see e.g. [1, 2]), and also for generation of gas-permeable [3] and ion-selective [4] membranes. In this work is given new data concerning synthesis of fluorinated norbornenes (1-3) and 7-oxanorbornene (4) (see scheme 1), which are prospective monomers for such polymersb $万^{\text {TM }}$ production.


It should be noted，that there are some single reports about present compounds in the literature． Thus，the hexafluorotricyclononene（1）obtaining was declared in the publication［5］，and its hydroformylation was reported about in the article［6］．But nowhere was given neither detailed description of the compound（1）synthesis，nor its characteristics．

We obtained tricyclononene（1）as the mixture of endo－and exo－isomers（16：1）by prolonged heating of cyclopentadiene（CPD）（5）with the excess of perfluorocyclobutene（PFCB）（6）in autoclave without solvent，and the yield amounted to $34 \%$ ．Similarly，by the heating of furan（7）with the excess of PFCB （6）was synthesized also 9－oxatricyclononene（4），isolated with the yield $23 \%$ ．Low yield of products may be explained by necessity of end tricyclononene（1）deep purification from dicyclopentadiene （DCPD）impurity in the first case，and in the second case the reason was in the following［2＋4］－ cycloaddition reaction of compound（4）with furan（7）in this conditions（NMR and GC－MS data were not given）．

The generation of 9 －oxatricyclononene（4）by heating to $50-60 \mathrm{~B}^{\circ} \mathrm{C}$ of ether solution of furan（7）and PFCB（6）equimolar mixture is described in the literature［7］．We didnB万 ${ }^{\text {TM }} \mathrm{t}$ succeed in replaying this experiment：the reaction almost failed by the temperature $50-100 \mathrm{~B}^{\circ} \mathrm{C}$ and only about $150 \mathrm{~B}^{\circ} \mathrm{C}$ the conversion of furan（7）became approximately $10 \%$ after 19 hours．Although the melting point and parameters of ${ }^{1} \mathrm{H}$ and ${ }^{19} \mathrm{~F}$ NMR spectra of our compound obtained are differing from literary data［7］， the totality of its spectral characteristics（NMR，Raman，MS，see Tables 1－3）allows us to assign it the ［2＋4］－cycloadduct structure（4）．

The obtaining of dispironorbornene（2）by reaction of CPD（5）with PFCB dimer（8），or with the mixture of PFCB dimers（8／9）in the CsF presence［8，9］is also described in the literature．So the authors claimed perfluorobicyclobutylidene（8）only to be able to undergo the Diels－Alder reaction， whereas the olefin（9）stayed inert，and only its isomerisation to dimer（8）occurred in presence of CsF ［9］．But we found in the result of more thorough purification and examination of reaction products，that heating of CPD（5）with the mixture of PFCB dimers（8／9）＊in the CsF presence，led to $[2+4]$－ cycloadducts from both olefins：dispironorbornene（2）and 2－nonafluorocyclobutyl substituted tricyclononene（3），herewith the yield of dispironorbornene（2）beeing always higher at any ratio of starting dimers（8／9）．Besides this it appeared that pure dispironorbornene（2）was solid，but not liquid as was reported in［9］＊＊．

Polymerization of synthesized fluorinated norbornenes $(\mathbf{1}, \mathbf{2})$ and also 7－oxanorbornene（4）was investigated in Philipps－University，Marburg，（FRG），by Prof．Heitz group［10］．

## Experimental

${ }^{1} \mathrm{H}$ ，and ${ }^{19} \mathrm{~F}$（with $\mathrm{CF}_{3} \mathrm{COOH}$ as the external standard）NMR spectra were recorded in $\mathrm{CDCl}_{3}$ on Bruker WP－200SY spectrometer．Raman spectra were detected on Ramanor HG－2S spectrometer． Mass and GC－MS spectra were recorded on VG 7070E and Polaris／GCQ mass spectrometers（EI， 70 eV ）．Preparative GLC（PGLC）was performed at $150 \mathrm{~B}^{\circ} \mathrm{C}$ on QF1 on chromosorb column（ $\mathrm{I}=4 \mathrm{~m}, \mathrm{~d}$ $=25 \mathrm{~mm}$ ）or on FS1265 on chromosorb column（ $\mathrm{I}=3.6 \mathrm{~m}, \mathrm{~d}=25 \mathrm{~mm}$ ）．

CPD（5）and furan（9）were obtained by pyrolysis of DCPD and pyromucic acid，accordingly，and kept at $-78 \mathrm{~B}^{\circ} \mathrm{C}$ ．PFCB（6）［11］and PFCB dimers（8／9）with monohydrocycloalkane（10）impurity［12］ were synthesized by known procedures，mixture ratio was evaluated according NMR spectra and GLC．

Synthesized compounds characteristics are represented in Tables 1－3．

## 2，3，3，4，4，5－Hexafluorotricyclo［4．2．1．0 ${ }^{2,5}$ ］nonene－7（1）

The mixture of $5.93 \mathrm{~g}(89.7 \mathrm{mmol})$ CPD（5）， 18.65 g （ 115.0 mmol ）PFCB（6）and 0.2 g hydroquinone was heated at $150 \mathrm{~B}^{\circ} \mathrm{C}$ for 72 h in autoclave．After cooling all volatile compounds were distilled off，the residue was filtered and autoclave was washed by small amount of ether． 11.94 g （58\％）of cycloadduct （1）was obtained by combined filtrates distillating，purity $\mathrm{B} \% \ltimes$ 「 $98 \%$ ，DCPD impurity $\sim 1.2 \%$（NMR，GLC）， b．p．58－59 $\mathrm{B}^{\circ} \mathrm{C}$（20 Torr）．PGLC（QF1）of distilled product gave 6.89 g （34\％）of tricyclononene（1）as colorless liquid，representing the mixture of endo／exo－isomers（16：1），purity $\mathrm{B} \% \stackrel{ }{ }$ Ґ $99.9 \%$ ，DCPD content B\％ $00.01 \%$（NMR，GLC）．

5－Norbornene－2－spiro－1B万 ${ }^{\text {TM }}$－$\left(2 \mathrm{~B} 万^{T M}, 2 \mathrm{~B} 万^{T M}, 3 \mathrm{~B} 万^{T M}, 3 \mathrm{~B} 万^{\mathrm{TM}}, 4 \mathrm{~B} 万^{T M}, 4 \mathrm{~B} 万^{T M}\right.$－ hexafluorocyclobutane）－3－spiro－1вЂќ－（2вЂќ，2вЂќ，Зв万ќ，ЗвЂќ，4в万ќ，4в万ќ－ hexafluorocyclobutane）
（2）and
2－heptafluorocyclobutyl－3，3，4，4，5－

The mixture of $1.78 \mathrm{~g}(27.0 \mathrm{mmol})$ CPD（5）， 6.84 g PFCB dimers（8／9）with monohydrocycloalkane （10）$\left[(8) /(9) /(10)=1.0 / 0.51 / 0.15 ; 20.7 \mathrm{mmol}\right.$ calculating on $\mathrm{C}_{8} \mathrm{~F}_{12}$ ］and $3.24 \mathrm{~g}(21.3 \mathrm{mmol})$ freshly calcined CsF were heated at $80 \mathrm{~B}^{\circ} \mathrm{C}$ for 16 h in periodically agitated sealed tube．After the reaction expired the supernatant liquid contained cycloadducts（2）and（3）in the ratio 4．4：1．0（NMR，GLC）．

Similarly， 3.19 g （ 48.3 mmol ）CPD（5）， 11.11 g PFCB dimers（8／9）with monohydrocycloalkane（10） $\left[(8) /(9) /(10)=1.0 / 0.82 / 0.33 ; 33.7 \mathrm{mmol}\right.$ calculating on $\left.\mathrm{C}_{8} \mathrm{~F}_{12}\right]$ and 7.37 g （ 48.5 mmol ）CsF，and also $0.45 \mathrm{~g}(6.82 \mathrm{mmol}) \mathrm{CPD}(5), 1.49 \mathrm{~g} \mathrm{PFCB}$ dimers（8／9）with monohydrocycloalkane（10）［（8）／（9）／（10）＝ $1.0 / 4.5 / 3.9 ; 4.39 \mathrm{mmol}$ calculating on $\left.\mathrm{C}_{8} \mathrm{~F}_{12}\right]$ and $1.18 \mathrm{~g}(7.76 \mathrm{mmol}) \mathrm{CsF}$ were heated for 6 h ．The final solutions contained cycloadducts（2） $\mathrm{Pë}(3)$ in the ratio $\sim 2: 1$（NMR，GLC）．

Reaction mixtures were filtered，the precipitate was washed with ether，combined filtrates were concentrated and distilled，the fraction with b．p．61－66 $\mathrm{B}^{\circ} \mathrm{C}$（5 Torr）was collected．Distilled product PGLC（FS1265）gave 12.83 g （56\％）of cycloadduct（2）as white solid，with purity $\mathrm{B} \%$ 厄 ${ }_{\mathrm{o}}{ }^{2} 99 \%$ ，DCPD content $\mathrm{B} \% \mathrm{om} 0.01 \%$（NMR，GLC），and also 2.04 g （9\％）cycloadduct（3）as white fusible solid，with purity в\％「99\％（NMR，GLC）．

## 2，3，3，4，4，5－Hexafluoro－9－oxatricyclo［4．2．1．0 ${ }^{2,5}$ ］nonene－7（4）

The mixture of $9.6 \mathrm{~g}(0.141 \mathrm{~mol})$ furan（7）and $44.7 \mathrm{~g}(0.276 \mathrm{~mol})$ PFCB（6）was heated at $150 \mathrm{~B}^{\circ} \mathrm{C}$ for 72 h in autoclave．After cooling all volatile compounds were distilled off，the residue was solved in ether and filtered．Similarly，the ether solution of cycloadduct（4）was obained from $4.4 \mathrm{~g}(0.065 \mathrm{~mol})$ of furan （7）and $32.4 \mathrm{~g}(0.200 \mathrm{~mol})$ PFCB（6）．The combined ether solutions were concentrated and the residue was distilled，given 11.0 g （23\％）
9－oxatricyclononene（4）as colorless liquid，hardened to white crystalline substance with purity в\％厄 ${ }^{\text {® }} 99 \%$ ．

Table 1．Boiling and melting points，Raman spectra，and elemental analysis data for the compounds obtained．

| Compound | B．p．$/ B^{\circ} \mathrm{C}$ <br> （p／Torr） <br> ［m．p．$/ \mathrm{B}^{\circ} \mathrm{C}$ ］ | Raman，$\mathrm{OS} / \mathrm{cm}^{\mathrm{B} \text { 万"1 }}$ | Found（\％） <br> Calculated（\％） |  |  | Molecular formula |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | C | H | F |  |
| 1 | 57.5 （20） | 1569.0 vs（C＝C） | $\begin{aligned} & \underline{47.3} \\ & 47.4 \end{aligned}$ | $\begin{aligned} & \underline{2.50} \\ & 2.65 \end{aligned}$ | $\begin{aligned} & \underline{50.2} \\ & 50.0 \end{aligned}$ | $\mathrm{C}_{9} \mathrm{H}_{6} \mathrm{~F}_{6}$ |
| 2 | $\begin{aligned} & 71.5-72(7) \\ & {[52-53]^{a}}^{2} \end{aligned}$ | － | － | － | － | － |
| 3 | ［31－33］ | 1569.2 m（C＝C） | $\begin{aligned} & \underline{40.0} \\ & 40.0 \end{aligned}$ | $\begin{aligned} & 1.64 \\ & 1.55 \end{aligned}$ | $\begin{aligned} & \underline{58.2} \\ & 58.4 \end{aligned}$ | $\mathrm{C}_{13} \mathrm{H}_{6} \mathrm{~F}_{12}$ |
| 4 | $\begin{aligned} & 68-69(20) \\ & {[50-51.5]^{b}} \end{aligned}$ | 1568.5 vs（C＝C） | $\begin{aligned} & \underline{41.9} \\ & 41.8 \end{aligned}$ | $\begin{aligned} & \underline{1.78} \\ & 1.75 \end{aligned}$ | $\begin{aligned} & \underline{49.7} \\ & 49.5 \end{aligned}$ | $\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{6} \mathrm{O}$ |

${ }^{a}$ Colorless liquid［9］．${ }^{b}$ M．p． $59 B^{\circ} \mathrm{C}$［7］．
Table 2．NMR spectra of the compounds obtained．

|  | ${ }^{1} \mathrm{H}$ | ${ }^{19} \mathrm{~F}$ |
| :---: | :---: | :---: |
| 1 <br> endo－ isomer | $\begin{aligned} & 1.71\left(\mathrm{dtt}, 1 \mathrm{H}, \mathrm{H}(4),{ }^{2} \mathrm{~J}_{\mathrm{H}(4), \mathrm{H}(3)}=\right. \\ & 11.6,{ }^{4} \mathrm{~J}_{\mathrm{H}(4), \mathrm{F}(1)}=6.8,{ }^{5} \mathrm{~J}_{\mathrm{H}(4), \mathrm{F}(2 \text { or } 3)} \\ & =2.1) ; 1.89\left(\mathrm{br} . \mathrm{d}, 1 \mathrm{H}, \mathrm{H}(3),{ }^{2} J_{\mathrm{H}(3), \mathrm{H}(4)}\right. \\ & =11.6) ; \mathrm{H}(3), \mathrm{H}(4) \mathrm{B} \mathrm{~B}^{"} \mathrm{AB}-\mathrm{system} ; \\ & 3.38(\mathrm{~s}, 2 \mathrm{H}, \mathrm{H}(2)) ; 6.54(\mathrm{~s}, 2 \mathrm{H}, \mathrm{H}(1)) \end{aligned}$ | －112．5（br．dt，2F，$F(1),{ }^{4} J_{F(1), H(4)}$ B\％$€$ <br> ${ }^{3} J_{F(1), F(2}$ or 3$)$ в $\left.\% € 6.5\right)$ ；－51．3 and -50.4 （both dm，both $2 F, F(2)$ and $F(3)$ ， AB －system， $\mathrm{J}_{\mathrm{AB}}=225$ ） |
| 1 exo－ isomer | 2.24 （br．d， $1 \mathrm{H}, \mathrm{H}(3),{ }^{2} \mathrm{~J}_{\mathrm{H}(3), \mathrm{H}(4)}=$ 9．8）； 2.39 （br．dm，1H，H（4），${ }^{2} J_{H(4), H(3)}$ ＝9．8）；H（3），H（4）в万＂AB－system； 3.22 （br．m，2H，H（2））； 6.18 （br．m，2H， H（1）） | $-110.2 \mathrm{~m}, 2 \mathrm{~F}, \mathrm{~F}(1)) ;-49.7 \text { and }-47.8$ （both dm，both $2 F, F(2) P$ ë $F(3), A B-$ system， $\left.J_{\mathrm{AB}}=225\right)$ |
| 3 | 1.72 （br．dm，1H，H（5），${ }^{2} J_{H(5), H(4)}=$ 11．6）； 2.16 （br．d， $1 \mathrm{H}, \mathrm{H}(4),{ }^{2} \mathrm{~J}_{\mathrm{H}(4), \mathrm{H}(5)}$ ＝11．6）；H（4），H（5）в万＂AB－system； 3.47 Pë 3.56 （both br．s，both 1H，H（2） Pë H（3））； 6.63 （br．s，2H，H（1）） | －101．5（br．d．d，1F，$F(1), J_{d}=51.5$ ， $J_{d}=28.6$ ）；－94．8（br．s，1F，F（2））； －55．5－－39．9（m，10F，five $\mathrm{CF}_{2^{-}}$ groups） |
| 4 | $\begin{aligned} & 5.28 \text { (br.s, } 2 \mathrm{H}, \mathrm{H}(2)) ; 6.84 \text { (br.s, } 2 \mathrm{H} \text {, } \\ & \mathrm{H}(1))^{P^{\circ}} \end{aligned}$ | $\begin{aligned} & -114.5(m, 2 F, F(1)) ;-52.5(m, 4 F, \\ & F(2,3))^{P^{\circ}} \end{aligned}$ |

$P^{\circ}$ For the ${ }^{1}$ H Pë ${ }^{19}$ F NMR spectra，cf．Ref．［7］．
Table 3．Mass spectra of the compounds obtained．

| Com－ pound | $m / z\left(I_{r e l} .(\%)\right)$ |
| :---: | :---: |
| 1 endo－ isomer | $\begin{aligned} & 228[\mathrm{M}]^{+}(8.7) ; 209[\mathrm{M} \mathrm{в万"} \mathrm{~F}]^{+} \text {(0.6); } 189\left[\mathrm{C}_{9} \mathrm{H}_{5} \mathrm{~F}_{4} \text { and/or } \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{6}\right]^{+}(1.7) ; 169 \\ & {\left[\mathrm{C}_{6} \mathrm{H}_{2} \mathrm{~F}_{5}\right]^{+}(2.6) ; 163\left[\mathrm{C}_{7} \mathrm{H}_{3} \mathrm{~F}_{4}\right]^{+}(3.0) ; 145\left[\mathrm{C}_{7} \mathrm{H}_{4} \mathrm{~F}_{3}\right]^{+}(3.5) ; 128\left[\mathrm{M} \mathrm{в万"} \mathrm{C}_{2} \mathrm{~F}_{4}\right]^{+}} \\ & (8.7) ; 127 \\ & {\left[\mathrm{M} \mathrm{в万"} \mathrm{C}_{2} \mathrm{HF}_{4}\right]^{+}(19.6) ; 109\left[\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{~F}\right]^{+}(5.2) ; 69\left[\mathrm{CF}_{3}\right]^{+}(6.5) ; 66\left[\mathrm{C}_{5} \mathrm{H}_{6}\right]^{+}(100) ; 57} \\ & {\left[\mathrm{C}_{3} \mathrm{H}_{2} \mathrm{~F}\right]^{+}(6.5) ; 51\left[\mathrm{CHF}_{2}\right]^{+}(6.5) ; 39\left[\mathrm{C}_{3} \mathrm{H}_{3}\right]^{+}(8.7)} \end{aligned}$ |
| 1 exo－ isomer | $228[\mathrm{M}]^{+}$（15．7）； 209 ［M в万＂ F$]^{+}$（1．1）； $189\left[\mathrm{C}_{9} \mathrm{H}_{5} \mathrm{~F}_{4} \text { and／or } \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{6}\right]^{+}$（2．4）； 177 $\left[\mathrm{M} \mathrm{в万"} \mathrm{CHF}_{2}\right]^{+}(4.8) ; 169\left[\mathrm{C}_{6} \mathrm{H}_{2} \mathrm{~F}_{5}\right]^{+}(4.1) ; 163\left[\mathrm{C}_{7} \mathrm{H}_{3} \mathrm{~F}_{4}\right]^{+}(4.3) ; 159\left[\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{~F}_{3}\right]^{+}$ （4．8）； $145\left[\mathrm{C}_{7} \mathrm{H}_{4} \mathrm{~F}_{3}\right]^{+}$（8．9）； $133\left[\mathrm{C}_{6} \mathrm{H}_{4} \mathrm{~F}_{3}\right]^{+}$（7．8）； $128\left[\mathrm{M} \mathrm{в万"} \mathrm{C}_{2} \mathrm{~F}_{4}\right]^{+}$（17．8）； 127 ［M в万＂ $\left.\mathrm{C}_{2} \mathrm{HF}_{4}\right]^{+}$（43．0）； $109\left[\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{~F}\right]^{+}$（8．7）； $77\left[\mathrm{C}_{3} \mathrm{H}_{3} \mathrm{~F}_{2}\right]^{+}$（8．5）； $69\left[\mathrm{CF}_{3}\right]^{+}$（8．7）； 66 $\left[\mathrm{C}_{5} \mathrm{H}_{6}\right]^{+}$（100）； $57\left[\mathrm{C}_{3} \mathrm{H}_{2} \mathrm{~F}\right]^{+}$（7．4）； $51\left[\mathrm{CHF}_{2}\right]^{+}$（7．4）； $39\left[\mathrm{C}_{3} \mathrm{H}_{3}\right]^{+}$（8．7） |
| 2 | $\begin{aligned} & 390[\mathrm{M}]^{+}(2.3) ; 351\left[\mathrm{C}_{13} \mathrm{H}_{5} \mathrm{~F}_{10} \text { and/or } \mathrm{C}_{10} \mathrm{H}_{3} \mathrm{~F}_{12}\right]^{+}(3.2) ; 331\left[\mathrm{C}_{13} \mathrm{H}_{4} \mathrm{~F}_{9}\right. \text { and/or } \\ & \left.\mathrm{C}_{10} \mathrm{H}_{2} \mathrm{~F}_{11}\right]^{+} \text {(5.6); } 311\left[\mathrm{C}_{13} \mathrm{H}_{3} \mathrm{~F}_{8} \text { and/or } \mathrm{C}_{10} \mathrm{HF}_{10}\right]^{+}(6.6) ; 290\left[\mathrm{M} \mathrm{в万"} \mathrm{C}_{2} \mathrm{~F}_{4}\right]^{+} \\ & (22.8) ; 289\left[\mathrm{M} \mathrm{в万"} \mathrm{C}_{2} \mathrm{HF}_{4}\right]^{+}(25.4) ; 275\left[\mathrm{C}_{7} \mathrm{HF}_{10}\right]^{+}(16.4) ; 269\left[\mathrm{C}_{11} \mathrm{H}_{4} \mathrm{~F}_{7}\right]^{+}(23.2) \text {; } \end{aligned}$ |


|  | $239\left[\mathrm{M} \mathrm{в万"} \mathrm{C}_{3} \mathrm{HF}_{6}\right]^{+}$（28．0）； $231\left[\mathrm{C}_{8} \mathrm{H}_{2} \mathrm{~F}_{7}\right]^{+}$（22．7）； 221 ［М в万＂ $\left.\mathrm{C}_{3} \mathrm{~F}_{7}\right]^{+}$（26．6）； 219 $\left[\mathrm{C}_{10} \mathrm{H}_{4} \mathrm{~F}_{5}\right]^{+}(41.4) ; 201\left[\mathrm{C}_{10} \mathrm{H}_{5} \mathrm{~F}_{4}\right]^{+}$（49．4）； $200\left[\mathrm{C}_{10} \mathrm{H}_{4} \mathrm{~F}_{4}\right]^{+}(33.9) ; 190\left[\mathrm{C}_{9} \mathrm{H}_{6} \mathrm{~F}_{4}\right]^{+}$ （30．9）； $189\left[\mathrm{C}_{9} \mathrm{H}_{5} \mathrm{~F}_{4}\right]^{+}$（36．4）； $170\left[\mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{5}\right]^{+}$（43．0）； $169\left[\mathrm{C}_{6} \mathrm{H}_{2} \mathrm{~F}_{5}\right]^{+}$（50．9）； 151 $\left[\mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{4}\right]^{+}(40.1) ; 66\left[\mathrm{C}_{5} \mathrm{H}_{6}\right]^{+}(100) ; 65\left[\mathrm{C}_{5} \mathrm{H}_{5}\right]^{+}(39.4)^{a}$ |
| :---: | :---: |
| 3 | $390[\mathrm{M}]^{+}$（6．1）； 371 ［M в万＂F］${ }^{+}$（2．5）； 351 ［ $\mathrm{C}_{13} \mathrm{H}_{5} \mathrm{~F}_{10}$ and／or $\left.\mathrm{C}_{10} \mathrm{H}_{3} \mathrm{~F}_{12}\right]^{+}$（4．6）； $331\left[\mathrm{C}_{13} \mathrm{H}_{4} \mathrm{~F}_{9} \text { and／or } \mathrm{C}_{10} \mathrm{H}_{2} \mathrm{~F}_{11}\right]^{+}$（5．8）； $311\left[\mathrm{C}_{13} \mathrm{H}_{3} \mathrm{~F}_{8} \text { and／or } \mathrm{C}_{10} \mathrm{HF}_{10}\right]^{+}$（6．1）； 301 $\left[\mathrm{C}_{12} \mathrm{H}_{5} \mathrm{~F}_{8}\right]^{+}$（6．8）； 290 ［М в万＂ $\left.\mathrm{C}_{2} \mathrm{~F}_{4}\right]^{+}$（17．2）； 289 ［М в万＂ $\left.\mathrm{C}_{2} \mathrm{HF}_{4}\right]^{+}$（27．6）； 269 $\left[\mathrm{C}_{11} \mathrm{H}_{4} \mathrm{~F}_{7}\right]^{+}$（20．2）； 239 ［ M в万＂ $\left.\mathrm{C}_{3} \mathrm{HF}_{6}\right]^{+}$（25．9）； $219\left[\mathrm{C}_{10} \mathrm{H}_{4} \mathrm{~F}_{5}\right]^{+}$（39．7）； 201 $\left[\mathrm{C}_{10} \mathrm{H}_{5} \mathrm{~F}_{4}\right]^{+}(44.0) ; 200\left[\mathrm{C}_{10} \mathrm{H}_{4} \mathrm{~F}_{4}\right]^{+}(32.7) ; 189\left[\mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{6}\right]^{+}(25.6) ; 170\left[\mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{5}\right]^{+}$ （46．0）； $169\left[\mathrm{C}_{6} \mathrm{H}_{2} \mathrm{~F}_{5}\right]^{+}$（46．6）； $151\left[\mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{4}\right]^{+}$（31．6）； $115\left[\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~F}_{2}\right]^{+}$（37．9）； 66 $\left[\mathrm{C}_{5} \mathrm{H}_{6}\right]^{+}(100) ; 65\left[\mathrm{C}_{5} \mathrm{H}_{5}\right]^{+}(44.8)$ |
| 4 | $191\left[\mathrm{C}_{8} \mathrm{H}_{3} \mathrm{~F}_{4} \mathrm{O} \text { and／or } \mathrm{C}_{5} \mathrm{HF}_{6} \mathrm{O}\right]^{+}(0.2)$ ； $181\left[\mathrm{C}_{7} \mathrm{H}_{2} \mathrm{~F}_{5}\right]^{+}(0.6)$ ； $163\left[\mathrm{C}_{4} \mathrm{HF}_{6}\right]^{+}$（3．4）； $132\left[\mathrm{C}_{3} \mathrm{HF}_{5}\right]^{+}$（4．3）； $114\left[\mathrm{C}_{3} \mathrm{H}_{2} \mathrm{~F}_{4}\right]^{+}$（3．0）； $101\left[\mathrm{C}_{2} \mathrm{HF}_{4}\right]^{+}$（4．8）； $93\left[\mathrm{C}_{3} \mathrm{~F}_{3}\right]^{+}$（3．6）； 75 $\left[\mathrm{C}_{3} \mathrm{HF}_{2}\right]^{+}$（4．8）； $68\left[\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}\right]^{+}$（100）； $57\left[\mathrm{C}_{3} \mathrm{H}_{2} \mathrm{~F}\right]^{+}$（3．4）； $51\left[\mathrm{CHF}_{2}\right]^{+}$（3．4）； 40 $\left[\mathrm{C}_{3} \mathrm{H}_{4}\right]^{+}$（6．1）； $39\left[\mathrm{C}_{3} \mathrm{H}_{3}\right]^{+}$（8．7）； $31\left[\mathrm{CF}^{+}\right.$（5．5）； $29[\mathrm{CHO}]^{+}$（6．9） |

${ }^{a}$ For the mass spectrum，cf．Ref．［9］．
＊The mixture of PFCB dimers（8／9）used in this work contained some amount of hydrofluorination product B 万＂ 1－hydro－1－nonafluorcyclobutyl－2，2，3，3，4，4－hexafluorocyclobutane（10），which quantatively underwent dehydrofluorination in presence of CsF to fluoroolefins（8／9）in situ
＊＊Parameters of ${ }^{1} \mathrm{H}$ and ${ }^{19}$ F NMR spectra of dispironorbornene（2）obtained here virtually coincide with literary data［9］．

## References

1．C．Chang，D．Barnes，L．D．Seger，L．F．Rhodes，R．P．Lattimer，G．M．Benedikt，J．Photopolym． Sci．Technol．，2012，25，161－169．
2．T．Yamashita，M．Morita，Y．Tanaka，J．J．Santillan，T．Itani，J．Photopolym．Sci．Technol．，2011，24， 165－172．
3．J．Vargas，A．A．Santiago，M．A．Tlenkopatchev，M．Lopez－Gonzalez，E．Riande，J．Membr．Sci．， 2010，361，78－88．
4．A．A．Santiago，J．Vargas，J．Cruz－Gomez，M．A．Tlenkopatchev，R．Gavino，M．Lopez－Gonzalez， E．Riande，Polymer，2011，52，4208－4220．
5．V．A．Alb万 ${ }^{\text {TM }}$ bekov，V．N．Mironov，A．F．Benda，P．F．Potashnikov，G．A．Sokolb万 ${ }^{\text {tm }}$ sky，Abstr．of V National Organofluorine Conference（May 20－22 1986），Moscow，1986，p． 3.
6．I．I．Kerov，V．A．Yashkir，G．A．Korneeva，I．I．Krylov，E．V．Slivinskii，A．V．Ignatenko，Petroleum Chemistry，1998，38，156－163（Engl．Transl．）．
7．V．A．Alb万™ bekov，A．F．Benda，A．F．GontarB万 ${ }^{\text {TM }}$ ，G．A．SokolB万 ${ }^{\text {TM }}$ sky，I．L．Knunyants，Bull． Acad．Sci．USSR，Div．Chem．Sci．，1988，Vol．37，No． 4 （Engl．Transl．）．
8．R．D．Chambers，J．R．Kirk，G．Taylor，R．L．Powell，J．Fluor．Chem．，1983，22，393－395．
9．A．E．Bayliff，M．R．Bryce，R．D．Chambers，J．R．Kirk，G．Taylor，J．Chem．Soc．，Perkin Trans．1， 1985，1191－1193．
10．T．Haselwander，Dissertation，Marburg， 1996.
11．Syntheses of Organofluorine Compounds，Eds．I．L．Knunyants，G．G．Yakobson，Khimiya， Moscow，1973，p． 46.
12．R．D．Chambers，G．Taylor，R．L．Powell，J．Chem．Soc．，Perkin Trans．1，1980，426－428．

