

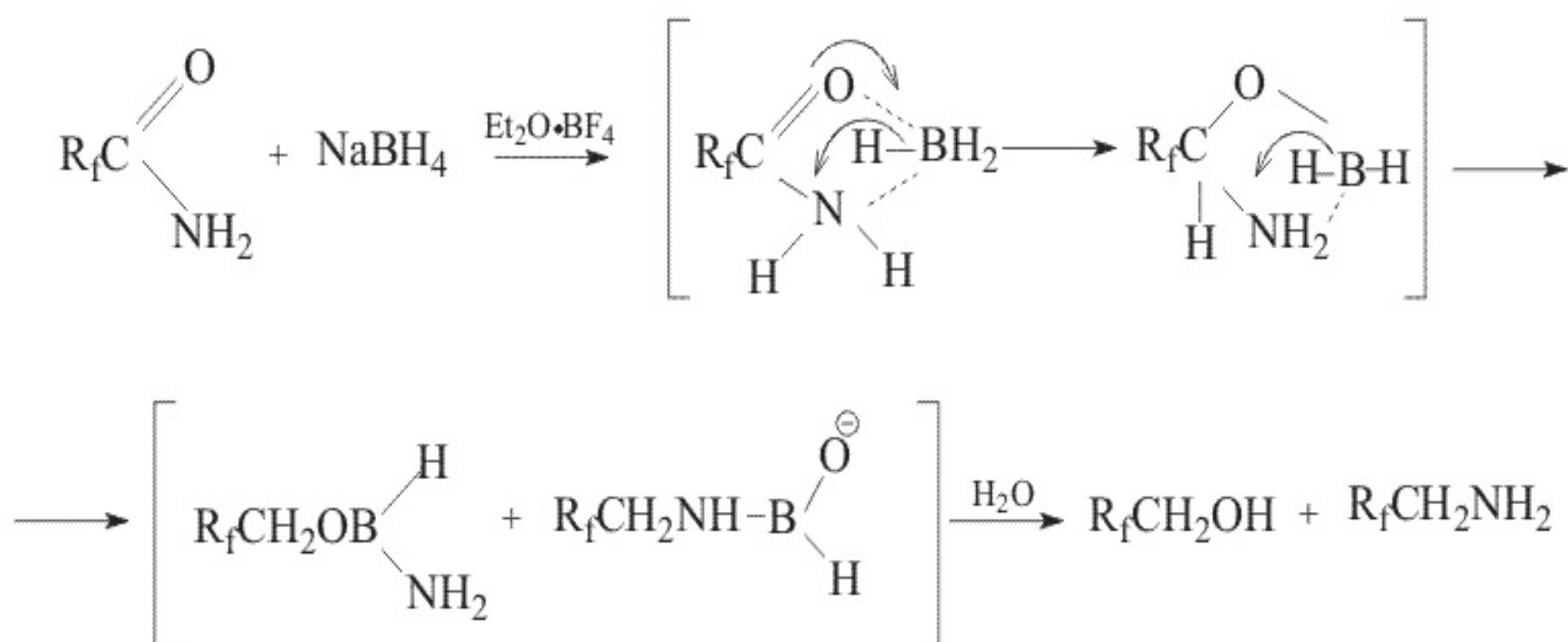
## 1,1-DIHYDROPERFLUOROALKYL AMINES. Repc

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Reduction of primary amides of acids with borane ( $\text{BH}_3$ ), generated in Situ by addition of  $\text{Et}_2\text{O}$  and  $\text{NaBH}_4$  in an aprotic solution is well known [1,2]. We have shown earlier [3] that the mixture of corresponding primary amine and alcohol.

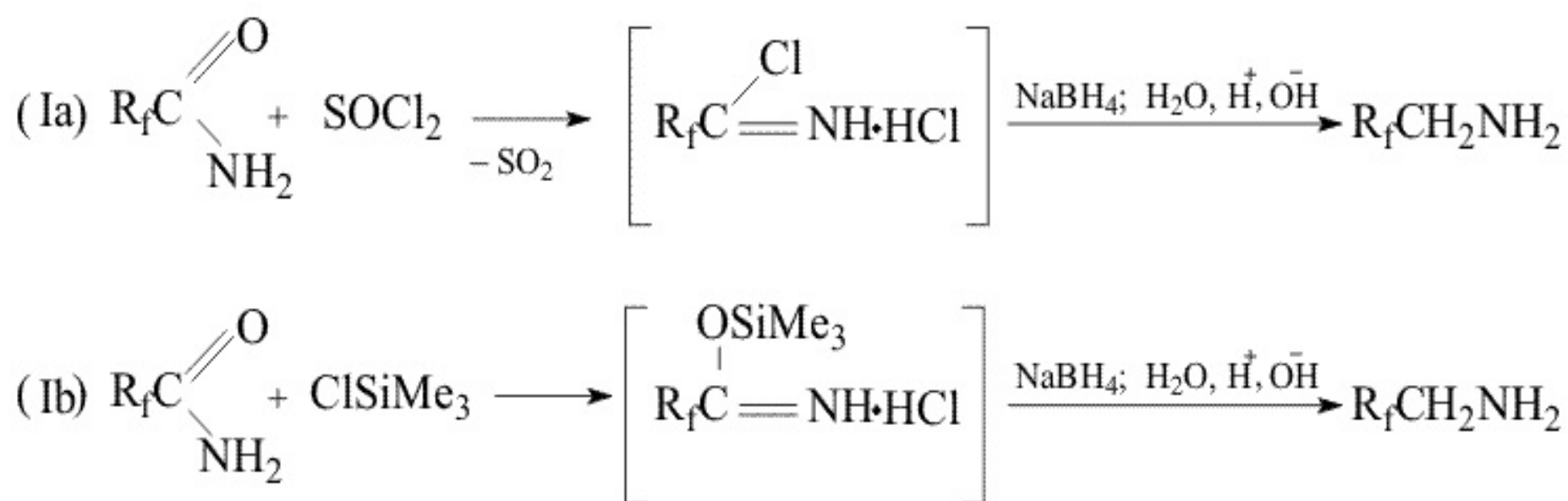


Throughout the research a new method has been developed, that makes possible the reduction of perfluoroacids to amines excluding formation of alcohols.

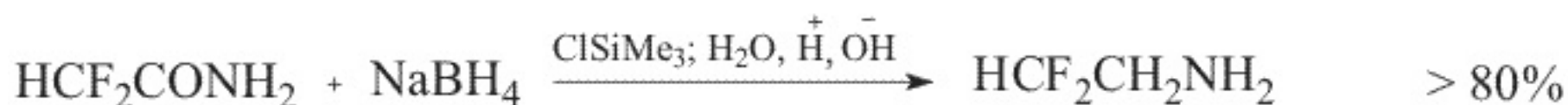
The method consists in treatment of the mixture of the amide and  $\text{NaBH}_4$  in an aprotic solvent like dichloromethane or chlorotrimethylsilane.

Probably in the first case ( $\text{SOCl}_2$ ) the formation of imidoylchloride takes place and in the second case of aminoacid is formed. As a result, in the first case oxygen is absent in the initial molecule that suppresses the process of alcohol formation, in the second case oxygen is bound with silicon that suppresses the process of alcohol formation.

The use of such a system results in a yield increase of the desired amine up to 80% and a suppression of the competitive reaction of the alcohol formation.



The latter variant was scaled by us in production of 1,1-dihydrodifluoroethylamine up to 10.0 t



## Experimental

### Method Ia

Sodium boron hydride (5.07g, 0.15 mol) was added to the solution of difluoroacetamide in dioxane at 20°C and thionyl chloride (18g, 0.15 mol) was added dropwise. The reaction mixture was heated up to 100°C and kept for 1 hour, then cooled to 50°C and the solution of NaOH (8g) was added dropwise (with simultaneous distillation of the desired product boiling before 70°C).

There was obtained 5.05 g of the distillate with the product content of 67% (Gas liquid chromatography).

### Method Ib

Sodium boron hydride (3 kg, 75 mol) was added at stirring to the solution of difluoroacetamide in dioxane (15.8 kg). Trimethylchlorosilane (7.6 kg, 68 mol) was added to the obtained suspension. The temperature of the reaction mixture was stable. Then the temperature was gradually increased. The reaction mixture was kept for another 5-6 hours, then cooled to 75-80°C and the solution of water was added at such a rate that the raw product would be distilled evenly. The product residue was removed from the reactor to 100-105°C.

There was obtained 12.4 kg of the raw product with the product content of ~35%. After rectification of the amine with BP = 68-63°C in 84% yield, 97% + purity.

## References

1. E.R. Bissell, M. Finger. J. Am. Chem. Soc., 24, 1959, 1256-1259.
2. S.E. Ellner, I.S. Wittman, W.I. Connich., J. Org. Chem., 30, 1965, 3645-3950.
3. A.F. Gontar, V.L. Don, E.V. Igoumnova, S.M. Igoumnov, 1,1-DIHYDROPERFLUOROALKYL AMINES

