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## Photocatalytic radical addition of 1,1,2,2-tetrafluoro-1-iodoethane to alkenes

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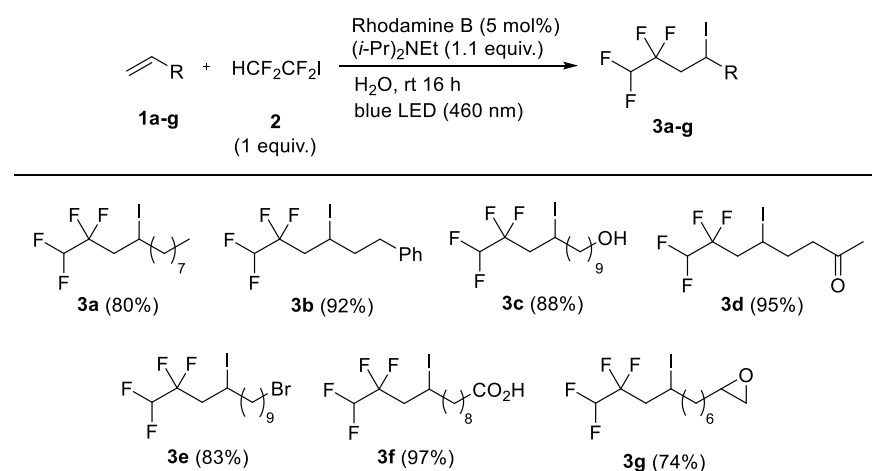
**Abstract:** Photocatalytic radical addition and addition-elimination reactions of terminal alkenes with 1,1,2,2-tetrafluoro-1-iodoethane under a blue LED light irradiation are reported. The photoredox catalysis with Rhodamine B or Eosin Y provided the corresponding 1,1,2,2-tetrafluoro-4-iodoalkanes or 1,1,2,2-tetrafluoroalkenes, respectively, in good yields. The obtained iododerivatives were successfully hydrodeiodinated to form saturated 1,1,2,2-tetrafluoroalkanes.

**Keywords:** photocatalysis; radical addition; Rhodamine B; Eosin Y; tetrafluoroiodoethane; LED

**Introduction:** Radical perfluoroalkylation is a popular method for the incorporation of perfluoroalkyl groups into organic molecules with a purpose of influencing properties of compounds [1-2]. Traditionally, perfluoroalkyl halides were used for the generation of perfluoroalkyl radicals; however, more recently metal sulfonates and sulfonyl chlorides have proven to be suitable perfluoroalkyl radical precursors [3-7]. Various approaches for homolytic cleavage of the carbon-halogen bond of perfluoroalkyl halides have been reported, these include radical initiators (AIBN [8], Et<sub>3</sub>B [9], or peroxide [10]), metals (magnesium [11], iron [1], or copper [12]), Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub> [13], catalytic triphenylphosphine [14], or UV light [15-16]. Recently, photoredox catalysis and in particular visible-light-promoted perfluoroalkylation [7, 17-20] have emerged as alternatives to initiator-promoted radical perfluoroalkylations. In this field, organic dyes enjoy an advantage over expensive ruthenium or iridium photocatalysts. Here we report Rhodamine B or Eosin Y catalyzed photocatalytic radical addition or addition-elimination, respectively, of HCF<sub>2</sub>CF<sub>2</sub>I to terminal alkenes. Compounds with the tetrafluoroethyl group are rare [21-26] and this work should improve access to them in order to study their properties.

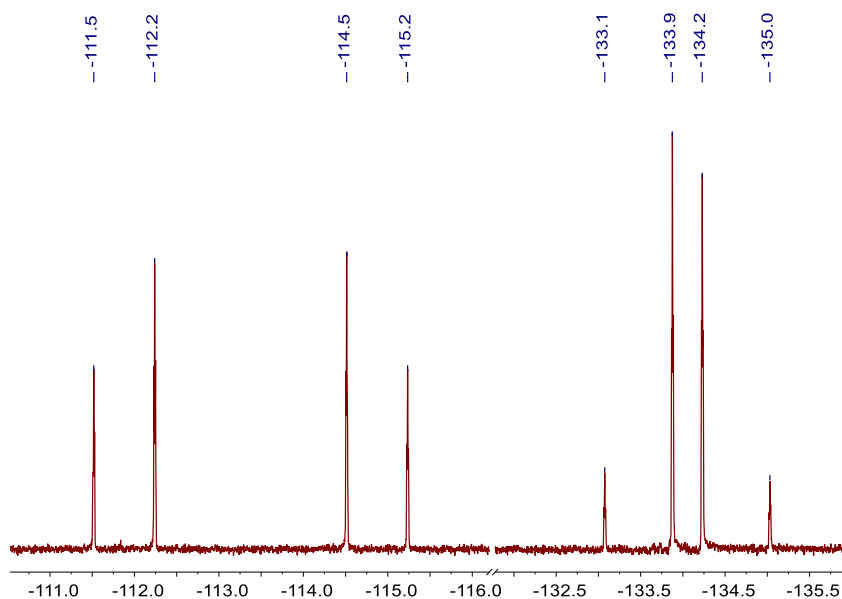
**Results and discussion:** In this work, we have used a blue LED light excited Rhodamine B [27] or Eosine Y [28] as photocatalysts in radical addition or addition-elimination reactions of HCF<sub>2</sub>CF<sub>2</sub>I with alkenes. Terminal alkenes **1** bearing different functional groups (alkyl, aryl, halogen, carbonyl, epoxy, hydroxyl, carboxylic acid) underwent radical addition of only equimolar amount of 1,1,2,2-tetrafluoro-1-iodoethane (**2**) in the presence of (*i*-Pr)<sub>2</sub>NEt and catalytic Rhodamine B in water to form the corresponding tetrafluoroiodoalkanes **3** in high yields (Scheme 1). Reactions proceeded smoothly at ambient temperature and complete conversion of **1** was reached under 16 hours (GC-MS control). Interestingly, an experiment with **3b** revealed that a green LED light (525 nm) could be used with similar efficiency to the blue light. Under the conditions shown in Scheme 1, hex-1-yne and *cis*-stilbene did not lead to the addition products.

### Scheme 1



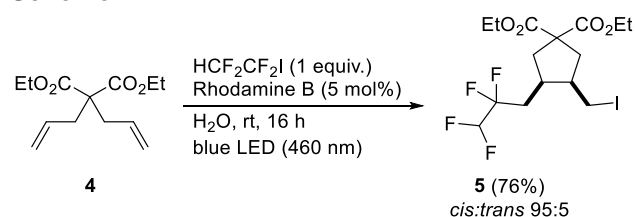
The  $^{19}\text{F}$  NMR spectra of iodo derivatives **3** showed an interesting splitting patterns owing to the presence of magnetically non-equivalent (diastereotopic) fluorine atoms (Figure 1).

**Figure 1**  $^{19}\text{F}$   $\{^1\text{H}\}$  NMR spectrum of **3a**



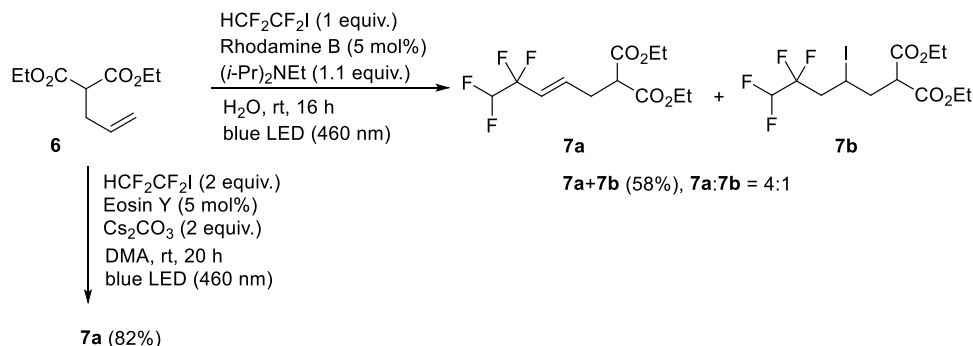
It can be expected that the mechanism operating involves a homolytic cleavage of the carbon-iodine bond and the formation of radicals which add to terminal alkenes. In order to probe this mechanism a reaction with diethyl 2,2-diallylmalonate (**4**) was performed. The formed fluoroalkyl radical induced a cascade radical addition – 5-exo-trig cyclization to the cyclopentane product **5**, which was isolated in good yield and high *cis*-selectivity (Scheme 2).

### Scheme 2



Diethyl allylmalonate (**6**) afforded under the Rhodamine B catalysis mostly alkene **7a** (product of HI elimination after radical addition) in a mixture with addition product **7b**. However, using Eosin Y as a photocatalyst, clean formation of **7a** was observed (Scheme 3). The formation of the elimination product is in an agreement with previous report using Eosin Y and cesium carbonate [28].

### Scheme 3



Next, we turned our attention to substitution of iodine atom of compounds **3** for hydrogen in order to access terminal tetrafluoroethyl-substituted aliphatic compounds. Towards this goal, compound **3b** was treated with excess  $\text{LiAlH}_4$  in THF to form (5,5,6,6-tetrafluorohexyl)benzene (**8**) in high yield (Scheme 4). This reaction was found to be substrate dependent: when epoxide **3g** was reduced either with  $\text{LiAlH}_4$  or with  $\text{H}_2/\text{Pd}$ , no reaction was observed even after prolonged reaction time (48 h) and the starting compound was isolated unchanged. The  $^{19}\text{F}$  NMR spectrum of **8** contains two set of signals: a doublet ( $^2J_{\text{FH}} = 54.0$  Hz) for the  $\text{CF}_2\text{H}$  group and a triplet of doublets ( $^3J_{\text{FH}} = 18.2, 2.0$  Hz) for the  $\text{CF}_2$  group (Figure 2).

### Scheme 4

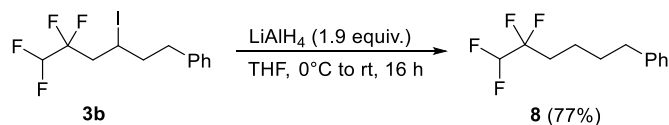
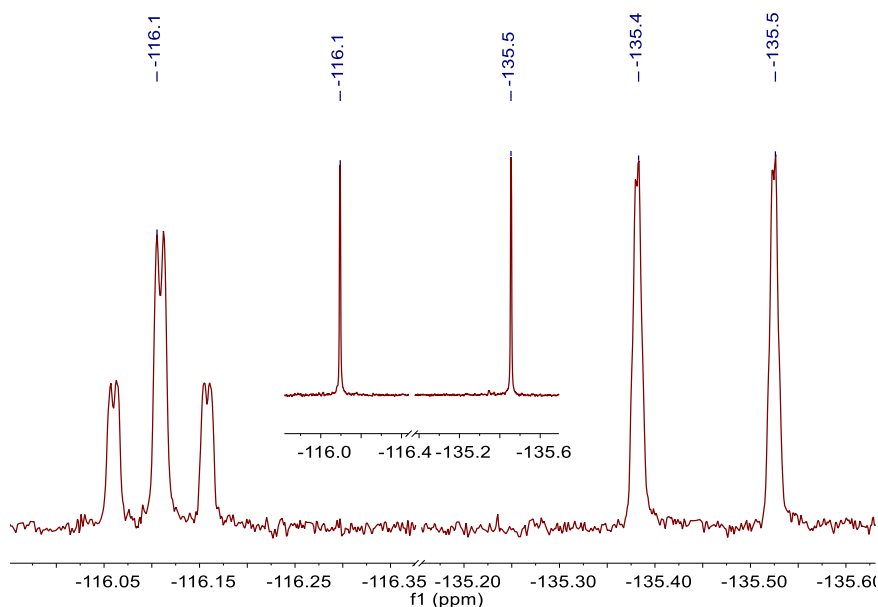


Figure 2.  $^{19}\text{F}$  NMR spectrum of **8**;  $^{19}\text{F}$   $\{^1\text{H}\}$  spectrum is shown as an insert



**Conclusion:** In conclusion, a combination of inexpensive organic dye catalyst and LED light radical addition of tetrafluoroiodoethane to terminal alkenes afforded 1,1,2,2-tetrafluoro-4-iodoalkanes in high yields. Alternatively, radical addition and subsequent HI elimination provided the corresponding 1,1,2,2-tetrafluoroalkenes. The iodo derivative was hydrodeiodinated with lithium aluminum hydride to form saturated 1,1,2,2-tetrafluoroalkane. Thus, tetrafluoroiodomethane ( $\text{HCF}_2\text{CF}_2\text{I}$ ) proved to be a useful and general reagent for radical introduction of the  $\text{HCF}_2\text{CF}_2$  group under photocatalytic conditions.

**Experimental part:** *General.* Reactions with air-sensitive materials were carried out under argon atmosphere using standard Schlenk techniques. All commercially available chemicals were used as received unless stated otherwise. Flash column chromatography was performed using silica gel 60 (0.040–0.063 mm).  $^1\text{H}$ ,  $^{13}\text{C}$ , and  $^{19}\text{F}$  NMR spectra were measured at ambient temperature using 5 mm diameter NMR tubes.  $^{13}\text{C}$  NMR spectra were proton decoupled. The chemical shift values ( $\delta$ ) are reported in ppm relative to internal  $\text{Me}_4\text{Si}$  (0 ppm for  $^1\text{H}$  and  $^{13}\text{C}$  NMR) or residual solvents and internal  $\text{CFCl}_3$  (0 ppm for  $^{19}\text{F}$  NMR). Coupling constants ( $J$ ) are reported in Hertz. GC-MS spectra were recorded using a 5% phenyl methylpolysiloxane column coupled with a quadrupole mass-selective electron impact (EI) detector (70 eV). High resolution MS spectra (HRMS) were recorded using electrospray (ESI), electron impact (EI) or dual ESI/MALDI ionizations. A 3 W LED sources (blue 455–475 nm or green 515–535 nm) containing a light-conductive glass rod immersed directly into the reaction vessel were used.

*Synthesis of compounds 3.* A mixture of alkene **1** (1.0 mmol) and water (10 mL) was cooled to 0 °C and degassed using three pump-thaw cycles under nitrogen atmosphere. To this emulsion were added (*i*-Pr) $_2$ NEt (192  $\mu\text{L}$ , 1.10 mmol), Rhodamine B (24 mg, 0.05 mmol), and  $\text{HCF}_2\text{CF}_2\text{I}$  (227 mg, 103  $\mu\text{L}$ , 1.0 mmol) using a pre-cooled syringe at 0 °C. The stirred reaction mixture was warmed to rt and irradiated with a blue LED lamp for 16 h (GC-MS control). The reaction mixture was then diluted with  $\text{Et}_2\text{O}$ , washed with water and brine. The organic layer was dried ( $\text{MgSO}_4$ ) and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (hexane:EtOAc, 97:3) to afford product **3**.

*1,1,2,2-Tetrafluoro-4-iodododecane (3a):* Yield: 80%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.72 (tdd,  $J = 53.9, 3.0, 2.3$  Hz, 1H), 4.39–4.27 (m, 1H), 2.96–2.57 (m, 2H), 1.91–1.69 (m, 2H), 1.61–1.38 (m, 2H), 1.36–1.21 (m, 10H), 0.89 (t,  $J = 7.1$  Hz, 3H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  117.52 (tt,  $J = 249.2, 29.4$  Hz), 110.03 (tt,  $J = 249.9, 41.1$  Hz), 40.85 (t,  $J = 21.2$  Hz), 40.64, 31.99, 29.73, 29.51, 29.37, 28.71, 22.81, 22.41 (t,  $J = 2.0$  Hz), 14.24;  $^{19}\text{F}$  NMR (377 MHz,  $\text{CDCl}_3$ )  $\delta$  –112.75 to –117.32 (m, 2F), –134.35 to –137.52 (m, 2F); HRMS (EI $^+$ )  $m/z$  calcd for  $\text{C}_{12}\text{H}_{21}\text{F}_4\text{I}$  [ $\text{M}$ ] $^+$ : 368.0624, found 368.0621.

*(5,5,6,6-Tetrafluoro-3-iodohexyl)benzene (3b):* Yield: 92%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35–7.26 (m, 2H), 7.27–7.17 (m, 3H), 5.70 (tdd,  $J = 53.9, 3.0, 2.2$  Hz, 1H), 4.28 (tt,  $J = 8.4, 5.3$  Hz, 1H), 3.00–2.63 (m, 4H), 2.13 (tt,  $J = 7.9, 5.0$  Hz, 2H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  140.25, 128.71, 128.66, 126.46, 117.50 (tt,  $J = 249.5, 29.6$  Hz), 109.97 (tt,  $J = 250.1, 41.1$  Hz), 42.06 (d,  $J = 2.1$  Hz), 40.85 (t,  $J = 21.2$  Hz), 35.84, 21.52 (t,  $J = 2.5$  Hz);  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  –111.96 to –116.66 (m, 2F), –133.82 to –136.55 (m, 2F); HRMS (EI $^+$ )  $m/z$  calcd for  $\text{C}_{12}\text{H}_{13}\text{F}_4\text{I}$  [ $\text{M}$ ] $^+$ : 359.9998, found 359.9994.

*12,12,13,13-Tetrafluoro-10-iodotridecan-1-ol (3c):* Yield: 88%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.71 (tt,  $J = 53.9, 2.7$  Hz, 1H), 4.31 (tt,  $J = 8.6, 5.1$  Hz, 1H), 3.61 (t,  $J = 6.2$  Hz, 2H), 2.89–2.58 (m, 2H), 2.02 (s, 1H), 1.84–1.70 (m, 2H), 1.59–1.49 (m, 2H), 1.40–1.21 (m, 12H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  117.46 (tt,  $J = 249.1, 29.4$  Hz), 109.96 (tt,  $J = 250.0, 41.0$  Hz), 63.07, 40.78 (t,  $J = 21.1$  Hz), 40.53 (d,  $J = 1.9$  Hz), 32.86, 29.64, 29.57, 29.46, 29.40, 28.60, 25.82, 22.39;  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  –113.20 to –117.20 (m), –134.18 to –137.30 (m); HRMS (ESI $^+$ )  $m/z$  calcd for  $\text{C}_{13}\text{H}_{24}\text{F}_4\text{IO}$  [ $\text{M}+\text{H}$ ] $^+$ : 399.0808, found 399.0811.

*7,7,8,8-Tetrafluoro-5-iodooctan-2-one (3d):* Yield: 95%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.74 (tdd,  $J = 53.8, 3.2, 2.2$  Hz, 1H), 4.49–4.23 (m, 1H), 2.99–2.55 (m, 4H), 2.17 (s, 3H), 2.16–2.07 (m, 1H), 1.97 (dddd,  $J = 15.2, 9.8, 8.1, 5.4$  Hz, 1H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  206.79, 117.34 (tt,  $J = 249.4, 29.4$  Hz), 109.92 (tt,  $J = 250.1, 40.8$  Hz), 43.75, 41.03 (t,  $J = 21.3$  Hz), 34.25 (d,  $J = 2.2$  Hz), 30.21,

21.21 (t,  $J = 2.7$  Hz);  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -112.26 to -116.45 (m), -134.10 to -136.77 (m); HRMS (ESI<sup>+</sup>)  $m/z$  calcd for  $\text{C}_8\text{H}_{12}\text{F}_4\text{IO}$  [M+H]<sup>+</sup>: 326.9869, found 326.9866.

**13-Bromo-1,1,2,2-tetrafluoro-4-iodotridecane (3e)**: Yield: 83%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.72 (tdd,  $J = 53.9, 3.1, 2.2$  Hz, 1H), 4.32 (tdd,  $J = 8.3, 5.8, 4.3$  Hz, 1H), 3.40 (t,  $J = 6.8$  Hz, 2H), 2.92–2.55 (m, 2H), 1.90–1.80 (m, 4H), 1.36–1.23 (m, 12H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  117.48 (tt,  $J = 249.1, 29.4$  Hz), 114.26, 109.98 (tt,  $J = 250.0, 41.0$  Hz), 40.81 (t,  $J = 21.2$  Hz), 40.54 (d,  $J = 1.9$  Hz), 34.11, 32.93, 29.65, 29.42, 29.37, 28.81, 28.59, 28.26, 22.37;  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -112.92 to -117.43 (m), -134.50 to -137.09 (m); HRMS (EI<sup>+</sup>)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{21}\text{F}_4\text{BrI}$  [M-H]<sup>+</sup>: 458.9808, found 368.9799.

**12,12,13,13-Tetrafluoro-10-iodotridecanoic acid (3f)**: Yield: 97%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.72 (tt,  $J = 53.9, 2.7$  Hz, 1H), 4.32 (tdd,  $J = 8.3, 5.7, 4.3$  Hz, 1H), 2.95–2.55 (m, 2H), 2.35 (t,  $J = 7.5$  Hz, 2H), 1.79 (ddp,  $J = 19.7, 9.8, 4.5$  Hz, 2H), 1.63 (p,  $J = 7.4$  Hz, 2H), 1.53 (tq,  $J = 7.7, 3.8, 2.8$  Hz, 1H), 1.42–1.29 (m, 9H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  180.37, 117.51 (tt,  $J = 249.1, 29.4$  Hz), 110.01 (tt,  $J = 250.1, 41.1$  Hz), 40.81 (t,  $J = 21.1$  Hz), 40.54 (d,  $J = 1.9$  Hz), 34.18, 29.65, 29.27, 29.24, 29.10, 28.57, 24.75, 22.37 (t,  $J = 2.5$  Hz);  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -112.70 to -117.55 (m), -134.26 to -137.53 (m); HRMS (ESI<sup>+</sup>)  $m/z$  calcd for  $\text{C}_{13}\text{H}_{22}\text{F}_4\text{IO}_2$  [M+H]<sup>+</sup>: 413.0601, found 413.0602.

**2-(9,9,10,10-Tetrafluoro-7-iododecyl)oxirane (3g)**: Yield: 74%; colorless oil,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.71 (tdd,  $J = 53.9, 3.2, 2.2$  Hz, 1H), 4.31 (tdd,  $J = 8.3, 5.7, 4.3$  Hz, 1H), 2.89 (tdd,  $J = 5.0, 3.9, 2.7$  Hz, 1H), 2.86–2.60 (m, 2H), 2.77–2.68 (m, 1H), 2.45 (dd,  $J = 5.0, 2.7$  Hz, 1H), 1.78 (ddt,  $J = 14.7, 9.6, 5.0$  Hz, 2H), 1.58–1.22 (m, 10H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  117.48 (tt,  $J = 249.2, 29.4$  Hz), 109.97 (tt,  $J = 250.1, 41.0$  Hz), 52.41, 47.18, 40.79 (t,  $J = 21.2$  Hz), 40.46 (d,  $J = 1.8$  Hz), 32.52, 29.57, 29.28, 28.54 (d,  $J = 1.8$  Hz), 25.99, 22.27 (t,  $J = 2.5$  Hz);  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -112.73 to -117.45 (m), -134.39 to -137.12 (m); HRMS (ESI<sup>+</sup>)  $m/z$  calcd for  $\text{C}_{12}\text{H}_{20}\text{F}_4\text{IO}$  [M+H]<sup>+</sup>: 383.0495, found 383.0493.

**Synthesis of compound 5, (diethyl cis-3-(iodomethyl)-4-(2,2,3,3-tetrafluoropropyl)cyclopentane-1,1-dicarboxylate)**: Diethyl 2,2-diallylmalonate (**4**) (120 mg, 0.5 mmol) and water (5 mL) were cooled to 0 °C and degassed using three pump–thaw cycles under nitrogen atmosphere. To this emulsion was added Rhodamine B (12 mg, 0.025 mmol) and then iodide **2** (227 mg, 103  $\mu\text{L}$ , 1.0 mmol) by pre-cooled syringe at 0 °C. The stirred reaction mixture was warmed to rt and irradiated with a blue LED lamp for 16 h (GC-MS control). The reaction mixture was then diluted with  $\text{Et}_2\text{O}$  and washed with water and brine. The organic phase was dried ( $\text{MgSO}_4$ ) and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (hexane:EtOAc, 99:1) to afford product **5** (180 mg of yellow oil, 76% yield) as the mixture of *cis:trans* isomers in ratio 95:5.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.70 (tt,  $J = 54.0, 2.6$  Hz, 1H), 5.11–5.07 (m, 1H), 5.06 (t,  $J = 1.2$  Hz, 1H), 4.26–4.08 (m, 4H), 3.18 (ddd,  $J = 9.8, 5.2, 1.0$  Hz, 1H), 3.01 (t,  $J = 9.9$  Hz, 1H), 2.61 (dt,  $J = 7.5, 1.2$  Hz, 2H), 2.57–2.43 (m, 2H), 2.34–2.18 (m, 1H), 2.14–1.82 (m, 1H), 1.28–1.15 (m, 6H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  172.03, 170.70, 132.30, 119.09, 117.85 (tt,  $J = 247.3, 29.5$  Hz), 110.11 (tt,  $J = 249.4, 41.6$  Hz), 61.76 (d,  $J = 9.6$  Hz), 61.19, 45.44, 39.73, 36.70, 35.41 (d,  $J = 2.2$  Hz), 28.45 (t,  $J = 21.8$  Hz), 14.09, 13.97 (d,  $J = 3.0$  Hz);  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -102.18 to -108.83 (m), -125.64 (ddd,  $J = 53.3, 38.9, 8.5$  Hz); HRMS (ESI<sup>+</sup>)  $m/z$  calcd for  $\text{C}_{15}\text{H}_{22}\text{F}_4\text{IO}_4$  [M+H]<sup>+</sup>: 469.04934, found 469.04935.

**Addition of  $\text{HCF}_2\text{CF}_2\text{I}$  to diethyl 2-allylmalonate 6**: a) **Reaction catalyzed with Rhodamine B**. A mixture of diethyl 2-allylmalonate (**6**) (200 mg, 1 mmol) and water (10 mL) was cooled to 0 °C and degassed using three pump–thaw cycles under nitrogen atmosphere. To this emulsion was added (*i*-Pr)<sub>2</sub>NEt (192  $\mu\text{L}$ , 1.10 mmol), Rhodamine B (24 mg, 0.050 mmol), and  $\text{HCF}_2\text{CF}_2\text{I}$  (227 mg, 103  $\mu\text{L}$ , 1.0 mmol) by pre-cooled syringe at 0 °C. The stirred reaction mixture was warmed to rt and irradiated with a blue LED lamp for 16 h. The reaction mixture contained according to the GC-MS analysis a mixture of the addition-elimination product **7a** and addition product **7b**. The dark-red suspension was diluted with  $\text{Et}_2\text{O}$ , washed with water and brine. The organic phase was dried ( $\text{MgSO}_4$ ) and then concentrated under reduced pressure. The residue was purified by silica gel column chromatography (hexane:EtOAc, 95:5) to afford a mixture **7a** and **7b** (250 mg of dark yellow oil, 58% yield, **7a:7b** 4:1). b) **Reaction catalyzed with Eosin Y**. A mixture of diethyl 2-allylmalonate (**6**) (200 mg, 1 mmol) and water (10 mL) was cooled to 0 °C and degassed using three pump–thaw cycles under nitrogen atmosphere. To this emulsion was added  $\text{Cs}_2\text{CO}_3$  (652 mg, 2 mmol), Eosin Y (32 mg, 0.05 mmol), and  $\text{HCF}_2\text{CF}_2\text{I}$  (454 mg, 206  $\mu\text{L}$ , 2.0

mmol) by pre-cooled syringe at 0 °C. The stirred reaction mixture was warmed to rt and irradiated with a blue LED lamp for 20 h. The reaction mixture contained according to the GC-MS analysis pure addition-elimination product **7a**. The red suspension was diluted with Et<sub>2</sub>O and washed with water and brine. The organic phase was dried (MgSO<sub>4</sub>) and concentrated under reduced pressure to yield 247 mg (82%) of colorless oil **7a**. *Diethyl (E)-2-(4,4,5,5-tetrafluoropent-2-en-1-yl)malonate (7a)*: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.75 (tt, *J* = 53.8, 2.9 Hz, 1H), 4.34–4.07 (m, 4H), 2.40–2.21 (m, 1H), 2.15–2.00 (m, 1H), 1.97–1.77 (m, 1H), 1.54–1.43 (m, 2H), 1.26 (td, *J* = 7.1, 5.5 Hz, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 169.47, 167.89, 117.38 (tt, *J* = 247.3, 29.4 Hz), 110.11 (tt, *J* = 249.4, 40.3 Hz), 61.87, 61.85, 32.99, 29.12 (t, *J* = 22.4 Hz), 20.25, 19.62 (t, *J* = 5.5 Hz), 14.15, 14.09; <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ –116.19 (t, *J* = 17.2 Hz), –136.03 (dd, *J* = 53.7, 26.3 Hz); HRMS (ESI<sup>+</sup>) *m/z* calcd for C<sub>12</sub>H<sub>17</sub>F<sub>4</sub>O<sub>4</sub> [M+H]<sup>+</sup>: 301.1063, found 301.1062.

*Reduction of (5,5,6,6-tetrafluoro-3-iodohexyl)benzene (3b)*. *(5,5,6,6-Tetrafluorohexyl)benzene (8)*: Compound **3b** (150 mg, 0.42 mmol) was dissolved in dry THF (2 mL) and cooled to 0 °C. LiAlH<sub>4</sub> (30 mg, 0.79 mmol) was then added and the reaction mixture was allowed to warm to rt and stirred for 16 h (GC-MS control). Saturated solution of NH<sub>4</sub>Cl (2 mL) and water (30 mL) were added and the reaction mixture was extracted with Et<sub>2</sub>O. The organic phase was dried (MgSO<sub>4</sub>) and then concentrated under reduced pressure. The residue was purified by silica gel column chromatography in hexane to afford pure **8** (75 mg of colorless oil, 77% yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.34–7.29 (m, 2H), 7.25–7.18 (m, 3H), 5.71 (tt, *J* = 54.1, 2.8 Hz, 1H), 2.67 (t, *J* = 7.5 Hz, 2H), 2.10–1.91 (m, 2H), 1.78–1.69 (m, 2H), 1.69–1.60 (m, 2H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 141.99, 128.53, 128.50, 126.04, 118.13 (tt, *J* = 245.8, 29.2 Hz), 110.47 (tt, *J* = 249.1, 41.5 Hz), 35.73, 31.22, 29.78 (t, *J* = 22.5 Hz), 20.20 (t, *J* = 3.8 Hz); <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ –116.11 (td, *J* = 18.2, 2.0 Hz), –135.45 (d, *J* = 54.0 Hz); HRMS (EI<sup>+</sup>) *m/z* calcd for C<sub>12</sub>H<sub>14</sub>F<sub>4</sub> [M]<sup>+</sup>: 234.1032, found 234.1033.

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